

This article was downloaded by:

On: 29 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713618290>

Conformations of Monosubstituted and Disubstituted 3,4'-,3,3'- and 4,4'-Diquinoliny Sulfides Studied by NMR Spectroscopy

Mirosław Wyszomirski^a; Andrzej Maślankiewicz^a; Krystian Pluta^a; Adolf Gogoll^b; Tadeusz Głowiak^c

^a Department of Organic Chemistry, Silesian School of Medicine, Sosnowiec, Poland ^b Department of Organic Chemistry, University of Uppsala, Uppsala, Sweden ^c Institute of Chemistry, University of Wrocław, Wrocław, Poland

To cite this Article Wyszomirski, Mirosław , Maślankiewicz, Andrzej , Pluta, Krystian , Gogoll, Adolf and Głowiak, Tadeusz(1994) 'Conformations of Monosubstituted and Disubstituted 3,4'-,3,3'- and 4,4'-Diquinoliny Sulfides Studied by NMR Spectroscopy', *Phosphorus, Sulfur, and Silicon and the Related Elements*, 95: 1, 415 — 416

To link to this Article: DOI: 10.1080/10426509408034257

URL: <http://dx.doi.org/10.1080/10426509408034257>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

CONFORMATIONS OF MONOSUBSTITUTED AND DISUBSTITUTED 3,4'-, 3,3'- AND 4,4'-DIQUINOLINYL SULFIDES STUDIED BY NMR SPECTROSCOPY

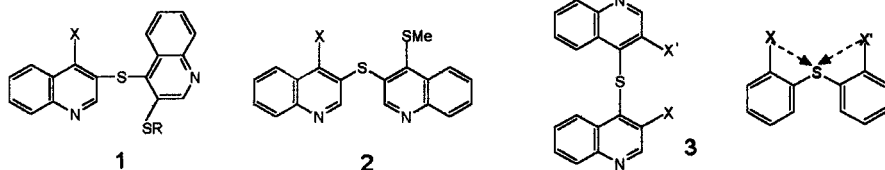
MIROSLAW WYSZOMIRSKI^a, ANDRZEJ MAŚLANKIEWICZ^a, KRYSTIAN PLUTA^a, ADOLF GOGOLL^b AND TADEUSZ GŁOWIAK^c

^a Department of Organic Chemistry, Silesian School of Medicine, Sosnowiec, Poland,

^b Department of Organic Chemistry, University of Uppsala, Uppsala, Sweden ,

^c Institute of Chemistry, University of Wrocław, Wrocław, Poland

3',4'-Disubstituted 3,4'-diquinolinyll sulfides **1** exhibit nontypical and unexpected spectral properties manifested mainly in proton NMR spectroscopy being results of different conformations at a central sulfide bridge¹. To complete studies on steric structure-spectral properties relationship in diquinolinyll sulfides several 4,4'-disubstituted-3,3'-diquinolinyll sulfides **2** and 3-mono- and 3,3'-disubstituted 4,4'-diquinolinyll sulfides **3** were synthesized and investigated by proton and carbon NMR spectroscopy .

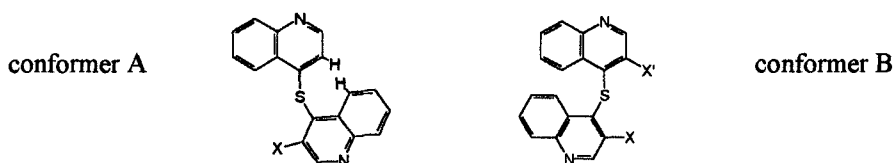


To verify our conclusion that the spectral properties of 3',4'-disubstituted 3,4'-diquinolinyll sulfides **1** are the final results of strong attractive interaction of both pairs of heteroatoms (as above), we started our study with proton and carbon NMR spectra of selected 4,4'-disubstituted 3,3'-diquinolinyll sulfides **2a-c**. In the case of unsymmetrical derivatives **2b** ($X = \text{OMe}$) and **2c** ($X = \text{Cl}$) significant differences in the chemical shift values of the H-2 and H-2' protons up to 0.55 ppm were observed. The unusual upfield shifts of the α -quinolinyll proton in sulfides **2b-c** were found for the H-2' proton located in methylthioquinoline moiety what is contrary to the results obtained for sulfides **1**². The H-2' hydrogen is located in aromatic shielding cone of the pyridine ring of methoxy- and chloroquinoline moieties. The differences in the chemical shifts of the C-2 and C-2' carbons (up to 8.1 ppm) seem to be more affected by the influence of the 4-heteroatom than by the confor-

mation of the whole molecule. Whereas the H-5 proton signal is shifted downfield (the through-space interactions with the 4-heteroatom) the C-5 carbon signal is found upfield (cf quinoline). The X-ray study confirmed a twist conformation for sulfide **2a** and a skew conformation for sulfide **2b**^{2,3}.

Although there was one conformer of dimethylthiocompounds **1a**⁴ and **2a**³ (X = SMe) in solid state, we observed two conformers for sulfide **3f** (X = H' = SMe)⁵. In proton and carbon spectra in CDCl₃ solution only one single structure was observed. For unsymmetrical sulfides **3b-c** (X = Cl, SMe, X' = H resp.) we observed the significant differences of the H-2 and H-2' proton signals (up to 0.61 ppm) and shielding of the H-3' proton signal (0.74 ppm).

The correlation peaks measured between the H-3' and H-5 protons in NOESY spectra and negative NOE enhancements in 1D ROESY spectra indicate the spatial proximity between the H-5 and H-3' protons in compounds **3a-b**.



To find out if there are structural preferences for any conformation in vacuo, extensive calculations were done using HyperChem v.3.0 package⁶. For compounds **3a-b** short distances between the H-5 and H-3' atoms were obtained, 3.56 Å and 3.34 Å. On the other hand, for compounds **3c** and **3f** these distances were longer than 5 Å, but short distances between the H-5 and H-5' atoms were found (2.83 Å and 2.61 Å resp.).

The obtained calculation results indicate the conformer A to be energetically favoured in vacuo for compounds **3a-b** and the conformer B for compounds **3c** and **3f**. These results are in accordance with NMR results for CDCl₃ solution.

REFERENCES

1. A. Maślankiewicz, M. Wyszomirski, S. Boryczka, A. Gogoll and T. Głowiak, *Phosphorus, Sulfur and Silicon*, **74**, 429, (1993).
2. K. Pluta and T. Głowiak, *J. Chem. Cryst.*, (in press).
3. K. Pluta, A. Maślankiewicz and T. Głowiak, *J. Cryst. Spectr. Res.*, **21**, 153 (1991).
4. A. Maślankiewicz, K. Pluta, S. Boryczka and T. Głowiak, *J. Cryst. Spectr. Res.*, **21**, 729 (1991).
5. K. Pluta, A. Maślankiewicz and T. Głowiak, *J. Cryst. Spectr. Res.*, **287**, (1993).
6. HyperChem v. 3.0, Autodesk, Inc., Sausalito, CA94965, USA